

I. Practice Problems**A. What is the mass of 1 mole (molar mass) of: Answers in g/mole**

- | | |
|-----------------------------------|---|
| 1. H_2 2.016 | 2. $\text{Mg}(\text{OH})_2$ 58.319 |
| 3. CO_2 44.009 | 4. NH_4Cl 53.492 |
| 5. CuSO_4 159.608 | 6. AgNO_3 169.874 |

B. Convert from grams to moles, or moles to grams

1. How many moles is 12.5 g of magnesium hydroxide?

.214 moles

2. How many moles is 1.46 g of hydrogen gas (
- H_2
-)?

.724 moles

3. How many grams are in 4.3 moles of ammonium chloride?

230 grams**C. Convert from moles to molecules, or molecules to moles**

1. How many molecules are in 2.0 moles of hydrogen gas (
- H_2
-)?

 1.2044×10^{24} molecules

2. How many moles is
- 2.0×10^{25}
- molecules of silver nitrate?

33.2 moles**D. How many atoms of oxygen are in 2.4×10^{23} molecules of copper(II) sulfate?** **9.6×10^{23} atoms of oxygen****II. Application Problems****A. How many molecules are in 96 g of carbon dioxide?** **1.31×10^{24} molecules****B. How many oxygen atoms are in 96 g of CO_2 ?** **2.627×10^{24} atoms****C. How many grams would 1.0×10^{25} molecules of copper(II) sulfate weigh?****2650 grams****D. How much does each individual molecule of copper(II) sulfate weigh?****159.608 amu's**

Mole Conversion Worksheet Answers

**Holt Rinehart & Winston, Holt,
Rinehart and Winston Staff**



Mole Conversion Worksheet Answers:

Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Chemistry 2e** Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course. The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning. The second edition has been revised to incorporate clearer, more current and more dynamic explanations while maintaining the same organization as the first edition. Substantial improvements have been made in the figures, illustrations and example exercises that support the text narrative. Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition. Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2012 Steve and Susan Zumdahl's texts focus on helping students build critical thinking skills through the process of becoming independent problem solvers. They help students learn to think like a chemist so they can apply the problem solving process to all aspects of their lives. In CHEMISTRY: AN ATOMS FIRST APPROACH 1e International Edition the Zumdahls use a meaningful approach that begins with the atom and proceeds through the concept of molecules, structure and bonding to more complex materials and their properties. Because this approach differs from what most students have experienced in high school courses it encourages them to focus on conceptual learning early in the course rather than relying on memorization and a plug and chug method of problem solving that even the best students can fall back on when confronted with familiar material. The atoms first organization provides an opportunity for students to use the tools of critical thinkers to ask questions, to apply rules and models and to Holt Chemistry Holt Rinehart & Winston, 2003-01-24

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Glencoe Chemistry: Matter and Change, Student Edition McGraw-Hill Education, 2016-06-15 **Comprehensive Organic Chemistry Experiments for the Laboratory Classroom** Carlos A. M. Afonso, Nuno R. Candeias, Dulce Pereira Simão, Alexandre F. Trindade, Jaime A. S. Coelho, Bin Tan, Robert Franzén, 2016-12-16 This expansive and practical textbook contains organic chemistry experiments for teaching in the laboratory at the undergraduate level covering a range of functional group transformations and key organic reactions The editorial team have collected contributions from around the world and standardized them for publication Each experiment will explore a modern chemistry scenario such as sustainable chemistry application in the pharmaceutical industry catalysis and material sciences to name a few All the experiments will be complemented with a set of questions to challenge the students and a section for the instructors concerning the results obtained and advice on getting the best outcome from the experiment A section covering practical aspects with tips and advice for the instructors together with the results obtained in the laboratory by students has been compiled for each experiment Targeted at professors and lecturers in chemistry this useful text will provide up to date experiments putting the science into context for the students *Quantities, Units and Symbols in Physical Chemistry* International Union of Pure and Applied Chemistry. Physical and Biophysical Chemistry Division, 2007 Prepared by the IUPAC Physical Chemistry Division

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